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Design and synthesis of 4",6"-unsaturated cyclic ADP-carbocyclic ribose as a Ca²⁺-mobilizing agent

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Abstract

4'',6''-Didehydro-cADPcR (3), an unsaturated carbocyclic ribose analog of a Ca²⁺-mobilizing second messenger cyclic ADP-ribose (cADPR), was designed and successfully synthesized using a key intramolecular condensation reaction forming the 18-membered pyrophosphate ring structure with a S-phenyl phosphorothioate-type substrate. Biological evaluation showed that 4'',6''-didehydro-cADPcR is a potent Ca²⁺-mobilizing agent in T cells.

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Cyclic ADP-ribose (cADPR, 1, Fig. 1) is a general mediator involved in Ca²⁺ signaling,¹ and synthesis of its analogs has been extensively investigated, since these can be used in investigating the mechanism of cADPR-mediated Ca²⁺ signaling pathways and are also expected to be leads for the development of potential drug candidates.^{2–5} In the synthesis of cADPR and its analogs, construction of the large 18-membered pyrophosphate ring structure is the key step, and we have developed an efficient method for forming the 18-membered ring by activating the *S*-phenyl phosphorothioate-type substrates with iodine or AgNO₃.⁴

cADPR is in a zwitterionic form with a positive charge around the $N(1)-C(6)-N^6$ moiety, making the molecule unstable, since the charged adenine moiety attached to the anomeric carbon of the N1-linked ribose can be an efficient leaving group to cleave the N1-ribosyl linkage. We previously designed and synthesized cyclic ADP-carbo-

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Fig. 1. cADPR (1), cADPcR (2), and 4",6"-didehydro-cADPcR (3).

cyclic-ribose (cADPcR, 2) as a stable mimic of cADPR, in which the oxygen atom in the N1-ribose ring of cADPR is replaced by a methylene group. cADPcR was shown to be actually resistant to both enzymatic and chemical hydrolysis, and it, like cADPR, effectively mobilizes intracellular Ca²⁺ in sea urchin eggs.^{4c}

Our attention has been focused on developing cell-type selective cADPR analogs, which can be useful as biological tools and/or potential drug leads. Thus, we found that although cADPcR and its 3'-deoxy derivative are more

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neplanocin A

Fig. 2. Naturally occurring carbocyclic adenine nucleosides.

HO

aristeromycin

potent than cADPR in neuronal cells,⁷ they are almost inactive in T cells, while cADPR works effectively as a Ca^{2+} -mobilizing second messenger both in T cell and in neuronal cell systems.^{5a-c} These very important findings indicate that the target proteins and/or the mechanism of action of cADPR in T cells and neuronal cells are different.^{5d}

Carbocyclic nucleosides are known as effective mimics of natural nucleosides.⁶ An antibiotic aristeromycin (Fig. 2) is a naturally occurring mimic of adenosine, 7 in which the ribose of adenosine is replaced by a carbocyclic-ribose, and has antiviral and antitumor activities, due to its competitive inhibitory effect to adenosine against S-adenosylhomocysteine (AdoHcy) hydrolase.⁸ Another nucleosidic antibiotic neplanocin A (Fig. 2), the structure of which corresponds to the 4'.6'-unsaturated derivative of aristeromycin, has significant antiviral and also antitumor effects, which clearly surpass those of aristeromycin.^{8,9} Therefore, since an unsaturated carbocyclic-ribose may mimic the ribose more precisely than the corresponding saturated congener, we newly designed the unsaturated carbocyclic ribose analog of cADPR, that is, 4".6"-didehvdro-cADPcR (3). We describe here synthesis and biological effects of 3 to show that it is significantly active in T cells.

We planned to synthesize the target 4",6"-didehydrocADPcR (3) by a route, as summarized in Scheme 1, in which intramolecular condensation reaction with a S-phenyl phosphorothioate-type substrate 4 forming the 18-membered pyrophosphate structure was employed. It was key point whether the condensation could occur, since the 18-membered pyrophosphate structure containing an unsaturated carbocyclic ribose seemed to be strained compared with the corresponding saturated congener. Substrate 4 could be converted from the N1-unsaturatedcarbocyclic-ribosyl adenosine derivative 5, which would be constructed by condensation between the optically active cyclopentenyl amine 6 with a known imidazole nucleoside 7.^{4c}

The chiral cyclopentenyl amine **6** was synthesized from a cyclopentenone **8**, which was prepared by the method reported by Jeong and co-workers¹⁰ (Scheme 2). 1,2-Reduction of the enone system of **8** with NaBH₄/CeCl₃ in MeOH stereoselectively gave the allylic α -alcohol **9**, of which mesylation and subsequent treatment with LiN₃ in HMPA/DMSO¹¹ afforded the β -azide **10**. Removal of the TBDPS group of **10** followed by reduction of the azido group with Ph₃P in aqueous THF gave the desired cyclopentenyl amine **6**.

The target 4".6"-didehydro-cADPcR (3) was successfully synthesized from the chiral cyclopentenyl amine 6 as shown in Scheme 3. The 4",6"-didehydro-N1-carbocyclicribosyl adenosine derivative 5 was obtained in 58% yield by the treatment of a mixture of the amine 6 and the imidazole nucleoside 7^{4c} with K₂CO₃ in MeOH at room temperature. The 5"-hydroxy group of 5 was protected with a dimethoxytrityl (DMTr) group, and the 5'-O-TBS group of the product was removed with TBAF to give 11. Treatment of 11 with an S,S'-diphenyl phosphorodithioate (PSS)/2,4,6-triisopropylbenzenesulfonyl chloride (TPSCl)/ pyridine system¹² gave the 5'-bis(phenylthio)phosphate 12. After removal of 5'-O-DMTr group of 12 with aqueous AcOH, a phosphoryl group was introduced at the resulting 5"-primary hydroxyl by Yoshikawa's method with POCl₃/ (EtO)₃PO,¹³ followed by treatment of the product with H_3PO_2 and Et_3N^{14} in the presence of N-methylmaleimide (NMM) in pyridine,^{4c} to afford the 5"-O-phosphoryl Sphenyl phosphorothioate 4, the substrate for the key intramolecular condensation.







Scheme 1.



Scheme 3. Reagents and conditions: (a) K_2CO_3 , MeOH, rt, 58%; (b) (1) DMTrCl, pyridine, rt; (2) TBAF, THF, AcOH, rt, quant; (c) PSS, TPSCI, pyridine, rt, 74%; (d) aq 60% AcOH, rt, 80%; (e) (1) POCl₃, (EtO)₃PO, 0 °C; (2) H₃PO₂, Et₃N, NMM, pyridine, 0 °C to rt, 28%; (f) AgNO₃, MS 3 Å, Et₃N, pyridine, rt, 65%; (g) aq HCO₂H, rt; (2) aq NH₃, 86%.

When a solution of **4** in pyridine was added slowly to a mixture of a large excess of AgNO₃ and Et₃N in the presence of MS 3 Å in pyridine at room temperature,^{4,15} the desired cyclization product **14** was successfully obtained in 65% yield, in spite of the strained 18-membered structure. Finally, removal of the isopropylidene groups of **14** with aqueous HCO₂H followed by treatment with aqueous ammonia furnished 4",6"-didehydro-cADPcR (**3**).¹⁷

The Ca²⁺-mobilizing effect of 4'',6''-didehydro-cADPcR (3) was evaluated in permeabilized Jurkat T cells. cADPR (1) released Ca²⁺ from intracellular stores as reported previously.^{3d} While cADPcR (2) as compared to cADPR (1) is



Fig. 3. Effects of cADPR (1), cADPcR (2), and 4",6"-didehydro-cADPcR (3) in permeabilized Jurkat T cells. Jurkat T cells were permeabilized by saponin. Ca²⁺ stores were refilled by addition of ATP and ATP-regenerating system. [Ca²⁺] was determined fluorimetrically using fura2/ free acid. Compounds were added at the concentrations indicated. Data are differences between the Ca²⁺ concentration upon addition and before addition of compound. Mean \pm SEM (n = 3-6).

only a weak agonist for Ca²⁺ release in permeabilized Jurkat T cells, 4",6"-didehydro-cADPcR (**3**) is much more potent than cADPcR (**2**) (Fig. 3). At 100 μ M 4",6"-didehydro-cADPcR induced a robust response while almost no effect of cADPcR was obtained. Interestingly, at 500 μ M 4",6"-didehydro-cADPcR a significantly higher amplitude was observed as compared to the natural second messenger cADPR. It is an important finding that, although the saturated carbocyclic analog cADPcR (**2**) is inactive, the corresponding 4",6"-unsaturated analog **3** has a significant Ca²⁺-mobilizing potency in T cells.

In conclusion, 4'',6''-didehydro-cADPcR (3) designed as a novel stable mimic of cADPR was successfully synthesized using the key intramolecular condensation reaction with the *S*-phenyl phosphorothioate-type substrate. Evaluation of Ca²⁺-mobilizing potency using permeabilized Jurkat T-lymphocytes showed that 4'',6''-didehydrocADPcR is significantly active in T cells and therefore it may be useful as a biological tool and/or drug lead because of the biologically and chemically stable carbocyclic structure.¹⁶

References and notes

- Cyclic ADP-ribose ADP-Ribose and NAADP: Structures, Metabolism and Functions; Lee, H. C., Ed.; Kluwer Academic: Dordrecht, 2002.
- Reviews on cADPR analogs: (a) Zhang, F.-J.; Gu, Q.-M.; Sih, C. J. Bioorg. Med. Chem. 1999, 7, 653–664; (b) Shuto, S.; Matsuda, A. Curr. Med. Chem. 2004, 11, 827–845; (c) Potter, B. V. L.; Walseth, T. F. Curr. Mol. Med. 2004, 4, 303–311.
- For examples: (a) Bailey, V. C.; Fortt, S. M.; Summerhill, R. J.; Galione, A.; Potter, B. V. L. *FEBS Lett.* **1996**, *379*, 227–230; (b) Galeone, A.; Mayol, L.; Oliviero, G.; Piccialli, G.; Varra, M. *Tetrahedron* **2002**, *58*, 363–368; (c) Xu, J.; Yang, Z.; Dammermann, W.; Zhang, L.; Guse, A. H.; Zhang, L.-H. J. Med. Chem. **2006**, 5501–

5512; (d) Wagner, G. K.; Guse, A. H.; Potter, B. V. L. J. Org. Chem. 2005, 70, 4810–4819.

- (a) Shuto, S.; Shirato, M.; Sumita, Y.; Ueno, Y.; Matsuda, A. J. Org. Chem. 1998, 63, 1986–1994; (b) Fukuoka, M.; Shuto, S.; Minakawa, N.; Ueno, Y.; Matsuda, A. J. Org. Chem. 2000, 65, 5238–5248; (c) Shuto, S.; Fukuoka, M.; Manikowsky, M.; Ueno, T.; Nakano, T.; Kuroda, R.; Kuroda, H.; Matsuda, A. J. Am. Chem. Soc. 2001, 123, 8750–8759.
- (a) Guse, A. H.; Cakir-Kiefer, C.; Fukuoka, M.; Shuto, S.; Weber, K.; Matsuda, A.; Mayer, G. W.; Oppenheimer, N.; Schuber, F.; Potter, B. V. L. Biochemistry 2002, 41, 6744–6751; (b) Shuto, S.; Fukuoka, M.; Kudoh, T.; Garnham, C.; Galione, A.; Potter, B. V. L.; Matsuda, A. J. Med. Chem. 2003, 46, 4741–4749; (c) Hashii, M.; Shuto, S.; Fukuoka, M.; Kudoh, T.; Matsuda, A.; Higashida, H. J. Neurochem. 2005, 94, 316–323; (d) Kudoh, T.; Fukuoka, M.; Ichikawa, S.; Murayama, T.; Ogawa, Y.; Hashii, M.; Higashida, H.; Kunerth, S.; Weber, K.; Guse, A. H.; Potter, B. V. L.; Matsuda, A.; Shuto, S. J. Am. Chem. Soc. 2005, 127, 8846–8855; (e) Kudoh, T.; Murayama, T.; Ogawa, Y.; Matsuda, A.; Shuto, S. Nucleosides, Nucleotides, Nucleic Acids 2006, 25, 583–599; (f) Kudoh, T.; Murayama, T.; Matsuda, A.; Shuto, S. Bioorg. Med. Chem. 2007, 15, 3032–3040.
- Agrofoglio, L. A.; Challand, S. R. Acyclic, Carbocyclic and L-Nucleosides; . Kluwer Academic: Dordrecht, 1998.
- Kusaka, T.; Yamamoto, H.; Shibata, M.; Muroi, M.; Kishi, T. J. Antibiot. 1968, 21, 255–263.
- 8. Wolfe, M. S.; Borchardt, R. T. J. Med. Chem. 1991, 34, 1521-1530.
- (a) Yaginuma, S.; Muto, N.; Tsujino, M.; Sudate, Y.; Hayashi, M.; Otani, M. . J. Antibiot. 1981, 34, 359–366; (b) Shuto, S.; Minakawa,

N.; Niizuma, S.; Kim, S.-H. ; Wataya, Y.; Matsuda, A. J. Med. Chem. 2002, 45, 748–751 and references cited therein.

- Choi, W. J.; Moon, H. R.; Kim, H. O.; Too, B. N.; Lee, J. A.; Shin, D. H.; Jeong, L. S. J. Org. Chem. 2004, 69, 2634–2636.
- Marquez, V. E.; Lim, M.-I.; Tseng, C. K.-H.; Markovac, A.; Priest, M. A.; Khan, M. S.; Kaskar, B. J. Org. Chem. 1988, 53, 5709–5714.
- (a) Sekine, M.; Nishiyama, S.; Kamimura, T.; Osaki, Y.; Hata, T. Bull. Chem. Soc. Jpn. 1985, 58, 850–860; (b) Sekine, M.; Hata, T. Cur. Org. Chem. 1993, 3, 25–66.
- Yoshikawa, M.; Kato, T.; Takenishi, T. Bull. Chem. Soc. Jpn. 1969, 42, 3505–3508.
- Hata, T.; Kamimura, T.; Urakami, K.; Kohno, K.; Sekine, M.; Kumagai, I.; Shinozaki, K.; Miura, K. Chem. Lett. 1987, 117–120.
- 15. The intermolecular condensation between a S-phenyl phosphorothioate and a phosphomonoester is effectively promoted by I_2 or AgNO₃ to give the corresponding pyrophosphate compound: see Ref. 12.
- In NG 108-15 neuronal cells,3 did not show apparent Ca²⁺mobilizing activity, which suggests that 3 is selectively active in T cells.
- 17. Physical data of **3** (triethylammonium salt): ¹H NMR (D₂O, 500 MHz, K⁺ salt) δ 8.63 (s, 1H), 8.49 (s, 1H), 6.31 (m, 1H), 6.07 (d, 1H, J = 5.7 Hz), 5.44 (m, 1H), 5.29 (m, 1H), 4.91 (m, 2H), 4.67 (m, 1H), 4.62 (m, 2H), 4.34 (m, 2H), 4.07 (m, 1H); ¹³C NMR (D₂O, 125 MHz) δ 153.1, 151.6, 147.4, 146.0, 143.3, 124.4, 121.0, 91.3, 85.0, 74.1, 73.6, 73.2, 71.0, 69.0, 64.7, 61.7; ³¹P NMR (D₂O, 202 MHz) δ -10.0 (d, J = 15.3 Hz), -10.6 (d, J = 15.3 Hz); HRMS (FAB, negative) calcd for C₁₆H₂₀N₅O₁₂P₂ 536.0589 [(M-H)⁻], found 536.0577; UV (H₂O) λ_{max} 260 nm.